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Key indicators

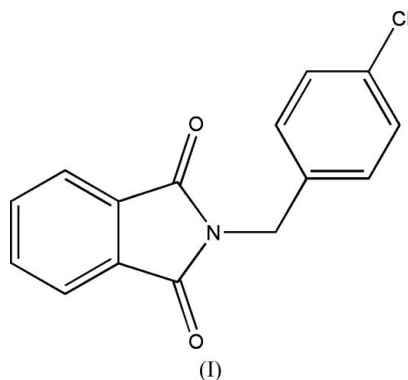
Single-crystal X-ray study
 $T = 294$ K
Mean $\sigma(\text{C}-\text{C}) = 0.005$ Å
 R factor = 0.051
 wR factor = 0.158
Data-to-parameter ratio = 14.8For details of how these key indicators were
automatically derived from the article, see
<http://journals.iucr.org/e>.*N*-(4-Chlorobenzyl)phthalimide

The title compound, $\text{C}_{15}\text{H}_{10}\text{ClNO}_2$, was synthesized by the reaction of 4-chlorobenzyl bromide with phthalimide. The phthalimide ring system is essentially planar and is twisted with respect to the chlorobenzene ring with a dihedral angle of $89.6(8)^\circ$.

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Comment

Phthalimide derivatives substituted by *N*-alkylation exhibit useful pharmaceutical properties (Chapman *et al.*, 1983; Donahoe *et al.*, 1957). The molecular structure of (I) is shown in Fig. 1. The phthalimide ring system is essentially planar, and twisted with respect to the C10–C15 benzene ring, with a dihedral angle of $89.6(8)^\circ$. In the crystal structure, molecules are linked *via* weak intermolecular C–H...O interactions (Table 1 and Fig. 2).



Experimental

Compound (I) was prepared according to the procedure reported by Cho *et al.* (1999). Phthalimide (1 g) in a solution in dimethylformamide (20 ml) was treated with potassium carbonate (0.94 g) at room temperature for 30 min. To the stirred solution, 4-chlorobenzyl bromide (1.39 g) was added and the mixture was stirred at room temperature for a further 8 h. The resulting mixture was poured into water (200 ml), yielding a white precipitate. The solid product was filtered off, washed with cold water and recrystallized from ethanol, giving single crystals of (I).

Crystal data

 $\text{C}_{15}\text{H}_{10}\text{ClNO}_2$
 $M_r = 271.69$
Monoclinic, $P2_1/c$
 $a = 12.796(5)$ Å
 $b = 14.005(5)$ Å
 $c = 7.188(3)$ Å
 $\beta = 103.763(6)^\circ$
 $V = 1251.2(8)$ Å³ $Z = 4$
 $D_x = 1.442$ Mg m⁻³
Mo $K\alpha$ radiation
 $\mu = 0.30$ mm⁻¹
 $T = 294(2)$ K
Block, colourless
 $0.22 \times 0.20 \times 0.14$ mm

Data collection

Bruker SMART CCD diffractometer	6736 measured reflections
φ and ω scans	2546 independent reflections
Absorption correction: multi-scan (SADABS; Bruker, 1997)	1102 reflections with $I > 2\sigma(I)$
$T_{\min} = 0.937$, $T_{\max} = 0.959$	$R_{\text{int}} = 0.054$
	$\theta_{\text{max}} = 26.4^\circ$

Refinement

Refinement on F^2	$w = 1/[\sigma^2(F_o^2) + (0.0661P)^2 + 0.1703P]$
$R[F^2 > 2\sigma(F^2)] = 0.051$	where $P = (F_o^2 + 2F_c^2)/3$
$wR(F^2) = 0.158$	$(\Delta/\sigma)_{\text{max}} < 0.001$
$S = 0.99$	$\Delta\rho_{\text{max}} = 0.24 \text{ e } \text{\AA}^{-3}$
2546 reflections	$\Delta\rho_{\text{min}} = -0.32 \text{ e } \text{\AA}^{-3}$
172 parameters	
H-atom parameters constrained	

Table 1Hydrogen-bond geometry (\AA , $^\circ$).

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
C3—H3 \cdots O2 ⁱ	0.93	2.54	3.461 (4)	169
C6—H6 \cdots O1 ⁱⁱ	0.93	2.41	3.314 (4)	165
C11—H11 \cdots O2 ⁱⁱⁱ	0.93	2.57	3.293 (4)	135

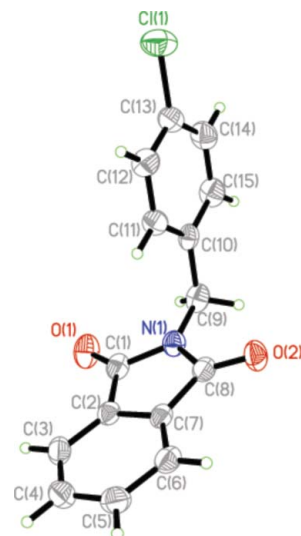
Symmetry codes: (i) $-x + 1, y + \frac{1}{2}, -z + \frac{3}{2}$; (ii) $-x + 1, y - \frac{1}{2}, -z + \frac{3}{2}$; (iii) $x, -y + \frac{1}{2}, z - \frac{1}{2}$.

H atoms were placed in calculated positions, with C—H = 0.93 (aromatic) and 0.97 \AA (methylene), and refined in the riding-model approximation, with $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{C})$.

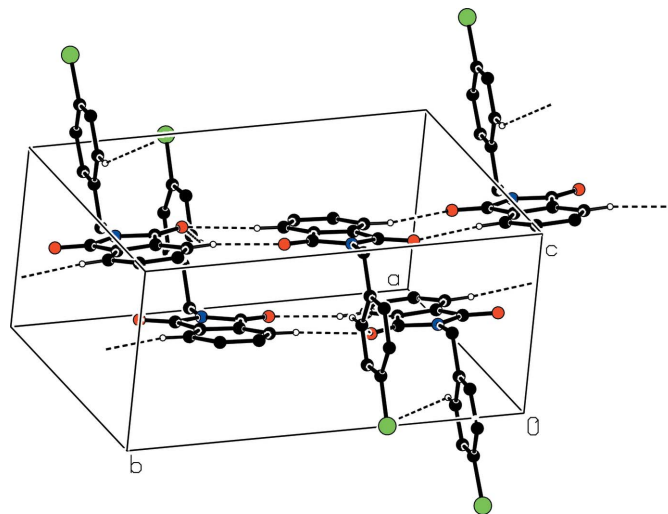
Data collection: *SMART* (Bruker, 1997); cell refinement: *SAINT* (Bruker, 1997); data reduction: *SAINT*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *SHELXTL* (Bruker, 1997) and *PLATON* (Spek, 2003); software used to prepare material for publication: *SHELXTL*.

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**Figure 1**

The molecular structure of (I), shown with 30% probability displacement ellipsoids (arbitrary spheres for H atoms).

**Figure 2**

Part of the crystal structure of (I), showing C—H \cdots O interactions as dashed lines.

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